

From atoms to steps: the microscopic origins of crystal growth.

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The Burton-Cabrera-Frank (BCF) theory of crystal growth has been successful in describing a wide range of phenomena in surface physics. Typical crystal surfaces are slightly mis-oriented with respect to a facet plane; thus, the BCF theory views such systems as composed of staircase-like structures of steps separating terraces. A continuous density of adsorbed atoms (adatoms) diffuses on the terraces, and steps move by absorbing or emitting flux of adatoms. In this paper, we shed light on the microscopic origins of the BCF theory by deriving a simple, 1D version of the theory from an atomistic, stochastic lattice-gas (SLG) model. We define the *time-dependent* adatom density and step position as appropriate ensemble averages in the SLG model, thereby exposing the non-equilibrium statistical mechanics origins of the BCF theory. Our analysis reveals that the BCF theory is valid in a *low adatom-density* regime, much in the same way that an ideal gas approximation applies to dilute gasses. We find conditions under which the surface remains in a low-density regime and discuss the microscopic origin of corrections to the BCF model.

I. INTRODUCTION

The controlled growth and etching of crystals is an important process that has applications in a variety of settings, including the fabrication of microprocessors, quantum dots, and nanowires, to name a few.⁸ Since these processes often involve the assembly of structures at the nanoscale, where the misplacement of a few atoms can have large effects, it is important to develop theoretical models that improve our understanding and control of such evolution processes. In particular, mesoscale models at the 10 nm – 100 μm range have gained considerable attention due to the fact that they provide a computationally tractable means to study discrete elements (e.g. defects) of nanoscale systems without tracking every individual atom.⁹ Formulating methods to connect atomistic and mesoscale models of crystalline surfaces is therefore a critical task in theoretical physics.

In this paper, we discuss the derivation of one such mesoscale model, the Burton-Cabrera-Frank (BCF) theory.¹⁰ In 1951, BCF postulated an important mechanism of surface evolution that came to be known as *step-flow*. They viewed crystal surfaces as composed of staircase-like structures, i.e. systems of steps separating terraces (cf. Fig. 1). Adsorbed atoms (adatoms) diffuse on the terraces until arriving at and attaching to a step. Such attachment (and the corresponding detachment) processes cause the steps to move, which can lead

to large scale morphological changes in the crystal over long times.

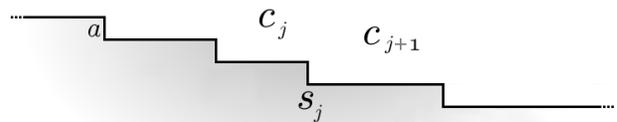


FIG. 1. A 1D step system with multiple steps (labeled s_j) separating terraces. Adatoms, represented by the densities c_j , diffuse on each terrace. The velocity of a step is proportional to the net current of adatoms arriving at the step.

Mathematically, Burton, Cabrera, and Frank chose to formulate this microscopic picture in terms of a Stefan-type free boundary problem.^{10,11} Adatoms are represented by a density c that obeys a diffusion equation; boundary conditions at the step account for the physics of attachment/detachment processes, and the velocity of a step, which is a free (or movable) boundary, is proportional to the net flux of adatoms arriving at the step.¹⁰ While this perspective is physically appealing, BCF did not derive their model from an atomistic theory of surface diffusion. Thus, many questions remain about the underlying assumptions of the theory. Is there an unambiguous relationship between the mesoscale parameters of the BCF theory the atomistic parameters describing

adatom motion? How far from equilibrium can a system be and still be well described by the BCF theory? How does the theory break down?

Our goal in this paper is to heuristically answer these questions by deriving a one-step, 1D version of the BCF theory from an idealized, atomistic model of surface diffusion. Our discussion here is largely formal; a full treatment of this problem requires a significant foray into set theory and functional analysis.¹ Nonetheless, a key observation allows for a simplification that makes the problem easily accessible: experiments and simulations show that the number of adatoms on many surfaces is typically very small. Thus, we choose as our starting point a *one-adatom* stochastic lattice-gas (SLG) master equation that describes the probability of finding the adatom at a particular location on the surface.

Our main task amounts to reconciling the stochastic and discrete nature of the SLG model with the notions of a deterministic and continuous adatom density and step position. This reconciliation can be achieved by defining the adatom density and step position as *time-dependent* expectation values taken over the master equation solution; the BCF theory, plus corrections, then describes the time evolution of these averages. This approach in particular has the benefit of showing how the BCF theory extends ideas of equilibrium statistical mechanics to non-equilibrium systems.

We caution, however, that the BCF theory (as we derive it) is not always valid out of equilibrium. Indeed, an important aspect of our analysis is to determine the “near-equilibrium” conditions under which the BCF theory reproduces the predictions of the SLG picture. To this end, we derive a maximum principle (often found in analysis of the heat equation)¹² to show when corrections to the BCF theory can be neglected. We also describe, but do not derive, corrections, due to *adatom correlations*, that arise from a multi-particle SLG model (as in Ref. 13) and discuss the conditions under which they can be neglected.

Several works have addressed questions related to the derivation of the BCF model.^{14–17} Here it suffices to note that these works only derived isolated parts of the BCF theory, whereas we aim to derive all of its elements together. However, the 1D, one-step SLG model that we invoke brings its own set of limitations. In particular, real crystal surfaces are 2-dimensional (2D), and steps usually have profiles that are not perfectly straight. These two facts greatly complicate the formulation of an atomistic model, and to the best of our knowledge, no derivation of a fully 2D BCF theory has been achieved. Nonetheless, we believe that our analysis sheds light on the key atomistic processes that give rise to the BCF theory.

The rest of the paper is organized as follows. In Sec. II, we present the mathematical elements of the BCF theory. In Sec. III, we present our one-step, 1D SLG model. In Sec. IV, we formulate an averaging procedure by which the SLG master equation can be transformed into discrete, BCF-type equations, plus corrections. In

Sec. V, we provide a maximum principle that yields conditions under which we can neglect the corrections, and in Sec. VI we take the continuum limit of the discrete BCF equations. In Sec. VII, we discuss corrections that arise from a multi-particle SLG model, and in Sec. VIII we present our main conclusions. Appendix A provides a proof of the maximum principle that we invoke in Section V.

II. A ONE-STEP BCF MODEL

We consider the one-step system illustrated in Fig. 2. A step at position $\zeta(t)$ separates an upper and a lower terrace. Adatoms, represented by the density $c(x, t)$, obey the diffusion equation

$$\partial_t c(x, t) = \mathcal{D} \partial_x^2 c(x, t), \quad 0 \leq x < \zeta(t), \quad \zeta(t) < x \leq L, \quad (1)$$

where \mathcal{D} is a constant diffusivity and L is the length of the system. We apply screw periodic boundary conditions at $x = 0$ and $x = L$.

Boundary conditions at the steps are of the form

$$\mathcal{J}_\pm = -\mathcal{D} \partial_x c|_\pm = \mp \kappa_\pm (c^\pm - c^{\text{eq}}), \quad (2)$$

where \mathcal{J}_\pm is the adatom flux at the right (+) or left(-) edge of the step, κ_\pm is an attachment/detachment rate at the right (+) or left (-) edge of the step,² and c^\pm is the adatom concentration to the right (+) or left (-) of the step; c^{eq} is an equilibrium adatom concentration. Equation (2) states that the step will emit or absorb adatoms until the densities c^\pm attain their equilibrium values. The density c^{eq} is generally assumed to be of the form

$$c^{\text{eq}} \propto e^{-\mu/k_B T}, \quad (3)$$

where μ is a *step chemical potential*, i.e. the energy added to the system when an atom attaches to the step.¹⁸

Because the step can move, we require an additional equation to describe its motion. We set the step velocity $\dot{\zeta}(t)$ equal to the net current

$$\dot{\zeta}(t) = a(\mathcal{J}_- - \mathcal{J}_+), \quad (4)$$

where a is the lattice spacing. Equation (4) is a mass conservation constraint; adatoms attaching to (detaching from) a step cause it to advance (retreat).

Our goal in the remainder of this paper is to derive Eqs. (1)–(4). In particular, an important part of our analysis is to identify κ_\pm and μ in terms of the processes in our SLG model.

III. STOCHASTIC LATTICE-GAS MODEL

In the context of surfaces, a SLG model is a probabilistic representation of the system that accounts for the random motion of individual atoms. Solutions to a

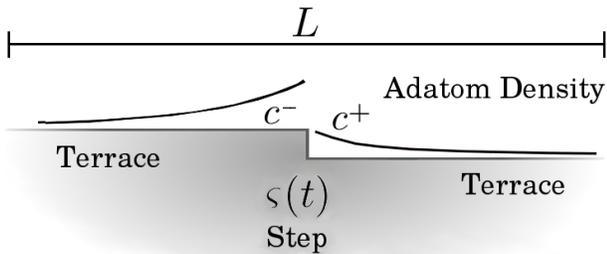


FIG. 2. The single-step system that we consider; the step position is denoted $\zeta(t)$. The values c^\pm are the adatom densities on the right (+) and left (-) sides of the step; L is the length of the system.

SLG model are the time-dependent probabilities of finding the system in each of its atomistic configurations. Given some initial state, the model describes how the system transitions between its accessible configurations.

In general, a SLG model allows for an arbitrary number of particles to move; see, for example, Refs. 13,17–19. However, in many experiments and simulations, one finds that few lattice sites are occupied by adatoms, which instead spend most of their time attached to a step. This observation motivates a key simplification of our SLG model: we only consider a system in which one adatom is ever allowed to move. While this simplification may seem drastic, Ref. 13 shows that, to good approximation, the behavior of a multi-particle model is often well described by the one-particle (1-p) model that we consider here. As we show in Sec. V, this 1-p model also contains the essential physics of the BCF theory.

Thus consider the system illustrated in Fig. 3. A surface with a single step is divided into N lattice sites indexed by j , where $0 \leq j \leq N - 1$. For definiteness, we pick a lattice site s_0 and use it to define a *microscopic* step position as follows:³ We call every site $j \neq s_0$ a “terrace site.” Whenever the adatom is at any site $j \neq s_0$, the *microscopic* step is at position $s_0 - 1$; when the adatom is at site s_0 , it becomes part of the step, whose position is then also s_0 .⁴ The SLG model is the following set of rules that describe how the atom hops between lattice sites.

Rule 1 *The adatom can only hop to one of its two adjacent lattice sites.*

Rule 2 *The adatom hops from a terrace site to any adjacent terrace site with a probability proportional to a constant rate D (described below).*

Rule 3 *The adatom hops to the step (i.e. $j = s_0$) from the left (-) or right (+) with probability proportional to an attachment rate $D\phi_\pm$ (defined below).*

Rule 4 *The adatom detaches from the step to the left (-) or right (+) with probability proportional to a detachment rate $Dk\phi_\pm$ (defined below).*

Analytically, these rules are expressed via a *master equation*, a system of ordinary differential equations describing the probabilities of finding the atom at each lattice site. If $p_j(t)$ is the probability that the atom is at site j , then the corresponding master equation is

$$\dot{p}_j = D[p_{j+1} - 2p_j + p_{j-1}], \quad j \neq 0, s_0, s_0 \pm 1, N-1, \quad (5)$$

$$\dot{p}_{s_0 \pm 1} = D[k\phi_\pm p_{s_0} - (1 + \phi_\pm)p_{s_0 \pm 1} + p_{s_0 \pm 2}], \quad (6)$$

$$\dot{p}_{s_0} = D[\phi_- p_{s_0-1} - k(\phi_- + \phi_+)p_{s_0} + \phi_+ p_{s_0+1}], \quad (7)$$

$$\dot{p}_0 = D[p_1 - 2p_0 + p_{N-1}], \quad (8)$$

$$\dot{p}_{N-1} = D[p_{N-2} - 2p_{N-1} + p_0], \quad (9)$$

where Eqs. (8) and (9) are screw periodic boundary conditions, e.g. when the adatom hops off of the right side of the system ($j = N - 1$) it reappears on the left side of the lattice (at $j = 0$). Equations (5), (8), and (9) encode Rule 2, while Eqs. (6) and (7) encode Rules 3 and 4, respectively. Rule 1 is expressed by each of Eqs. (5)–(9), since $\dot{p}_j(t)$ only depends on $p_j(t)$ and $p_{j\pm 1}(t)$.

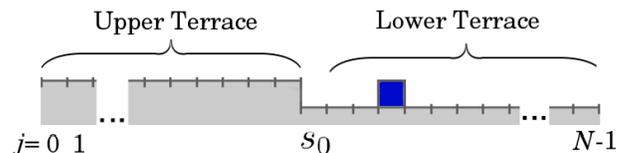


FIG. 3. The 1-p SLG model. A single atom (blue) is allowed to hop on the surface, whose lattice sites are indexed by j , where $0 \leq j \leq N - 1$. When the atom is at site s_0 it is a part of the step; otherwise it is an adatom.

The parameters D , ϕ_\pm , and k are often expressed as Arrhenius functions of the temperature.^{18,20} Specifically,

$$D = \tau^{-1} e^{-E_h/k_B T}, \quad (10)$$

$$\phi_\pm = e^{-E_\pm/k_B T}, \quad (11)$$

$$k = e^{-E_b/k_B T}, \quad (12)$$

where $k_B T$ is the temperature in units of energy, τ^{-1} is a hopping frequency that is usually assumed to be 10^{13} s^{-1} , E_h is an energy barrier to adatom hopping, E_\pm is an attachment barrier from the left (-) or right (+) of the step,⁵ and E_b is a bond energy (i.e. the energy increase of the system when an adatom detaches from the step). Physically, Eqs. (10)–(12) arise from the idea that an adatom must overcome an energy barrier in order to move to a new lattice site. In particular, an adatom that attaches to a step may form a bond that must be broken in a subsequent detachment process, while the barriers E_\pm account for the idea that adatom motion can be hindered when attaching to a step from above or below.

Equations (5)–(9) are supplemented by initial data satisfying the condition $\sum_j p_j(0) = 1$; i.e. there is unit probability of finding the particle somewhere on the surface. Summing Eqs. (5)–(9) then implies that $\sum_j p_j(t) = 1$ for all times (probability is conserved). It is also possible to show that (i) the system satisfies ergodicity (any configuration is accessible to any other configuration in a finite

number of transitions), (ii) there is a unique solution for any real initial data, and (iii) all initial data converges to the same steady state solution in the long-time limit. See Ref. 13 for a proof of statements (i)–(iii). It is also possible to solve Eqs. (5)–(9) *exactly* for all t , although we do not pursue this goal further. In the next section, we propose a procedure for averaging the master equation in such a way that yields the BCF model.

IV. AVERAGING THE SLG MODEL

Our goal in this section is to develop a suitable procedure by which we can transform the SLG master equation into a form resembling Eqs. (1)–(4). We must reconcile two differences between the SLG model and the BCF theory: (i) the adatom and step positions are represented as discrete quantities in the SLG model but continuous variables in the BCF theory, and (ii) coordinates on the terrace are represented by a discrete index j in the SLG model but a continuous variable x in the BCF theory. In this section, we address the first of these differences.

In statistical mechanics, measurable quantities are often defined as expectation values taken over an appropriate probability distribution, e.g. the Boltzmann distribution. Importantly, the expectation value of a random variable can be a continuous quantity, even though the random variable itself may only take discrete values. We employ this idea in defining the step position and adatom density of the BCF theory.

We begin by noting that $\dot{p}_j = 0$ if $p_j = k/Z$ for $j \neq s_0$ and $p_{s_0} = 1/Z$, where $Z = [1 + (N - 1)k]$ is a normalization constant. In light of Eq. (12), we conclude that this solution is in fact the Boltzmann distribution, which we now denote p_j^{eq} ; consequently, Z is the partition function. Since the steady state is unique, the system always approaches equilibrium at long times.

These observations motivate us to define *time-dependent* expectation values $\zeta(t)$ and $c_j(t)$ of the microscopic step position and adatom number-density at site j :

$$\zeta(t) = \left[\sum_{j \neq s_0} a(s_0 - 1)p_j(t) \right] + a s_0 p_{s_0}(t), \quad (13)$$

$$c_j(t) = p_j(t)/a \quad j \neq s_0, \quad (14)$$

where $a = L/N$, and L is the length of the system. As $t \rightarrow \infty$, the step position and adatom densities converge to their respective equilibrium expectation values, so that definitions (13) and (14) extend the notion of ensemble averaging to out-of-equilibrium systems.

If we apply a time derivative to Eqs. (13) and (14) and use Eqs. (5)–(9) to simplify the resulting expressions, we find the (discrete) step velocity law,

$$\begin{aligned} \dot{\zeta}(t) &= a^2 D \phi_- (c_{s_0-1} - k p_{s_0}/a) \\ &+ a^2 D \phi_+ (c_{s_0+1} - k p_{s_0}/a), \end{aligned} \quad (15)$$

and discrete diffusion-type equation,

$$\dot{c}_j(t) = D[c_{j-1} - 2c_j + c_{j+1}], \quad j \neq s_0, s_0 \pm 1, \quad (16)$$

$$\dot{c}_{s_0 \pm 1} = D[k\phi_{\pm} c_{s_0} - (\phi_{\pm} + 1)c_{s_0 \pm 1} + c_{s_0 \pm 2}], \quad (17)$$

$$\dot{c}_{s_0} = D[\phi_+ c_{s_0+1} - 2k(\phi_+ + \phi_-)c_{s_0} + \phi_- c_{s_0-1}]. \quad (18)$$

Equation (16) already has the form of Eq. (1) if we identify ∂_x^2 with the second-order difference scheme. However, Eqs. (17) and (18) do not have the same structure as Eq. (16); this fact allows us to determine boundary conditions for $c_j(t)$ corresponding to Eq. (2). Specifically, we add and subtract a new quantity $Dc_{s_0}^{\pm}$ to Eq. (17) to force the appearance of a second-order difference scheme plus some corrections, which we then require to vanish. This procedure may be interpreted as picking the boundary conditions for $c_1(t)$ such that the discrete diffusion-type equation (16) is valid all the way up to the step. Physically, then, we identify $c_{s_0}^{\pm}$ as the discrete analogues of c^{\pm} appearing in Eq. (2)

Following through with this procedure yields

$$\begin{aligned} \dot{c}_{s_0 \pm 1} &= D(c_{s_0}^{\pm} - 2c_{s_0 \pm 1} + c_{s_0 \pm 2}) \\ &+ D[(1 - \phi_{\pm})c_{s_0 \pm 1} + (k\phi_{\pm})p_{s_0}/a - c_{s_0}^{\pm}], \end{aligned} \quad (19)$$

where we treat the second line as the remainder term. Setting this equal to zero yields the discrete kinetic relations

$$\mp J_{\pm} = aD[c_{s_0 \pm 1} - c_{s_0}^{\pm}] = D\phi_{\pm}[c_{s_0 \pm 1} - k p_{s_0}/a], \quad (20)$$

where we identify $J_{\pm} = \pm(a^2 D)[c_{s_0 \pm 1} - c_{s_0}^{\pm}]/a$ as the discrete flux of adatoms on the right (+) and left (-) of the step. Note that J_{\pm} corresponds to $\mathcal{J}_{\pm} = -D\partial_x c(x, t)$, but with a first order difference scheme instead of a partial derivative in x .

V. MAXIMUM PRINCIPLE

At this point, we have all of the essential ingredients from which to derive the BCF theory: a step velocity law (15), a discrete diffusion equation (16), and discrete kinetic relations (20). It is tempting to take the continuum limit, but we must first acknowledge that Eq. (20) has no (constant) term corresponding to c^{eq} ; the only possible candidate is $k p_{s_0}/a$, which is a function of time.

This observation motivates the following idea: if p_{s_0} remains approximately constant for all times, then to good approximation, the term $k p_{s_0}$ in Eq. (20) can be replaced with a constant that we identify as the discrete analogue of c^{eq} in the linear kinetic relation [Eq. (2)]. To this end, we invoke a *maximum principle* (cf. Appendix A):

Proposition. *Let $p_j(t)$ be the solution to (5)–(9) with initial data $p_j(0)$, and define $\hat{p}_j = p_j/k$ for $j \neq s_0$ and $\hat{p}_{s_0} = p_{s_0}$. Then the greatest value in the set $\{\hat{p}_j(t)\}$ is less than or equal to the greatest value in the set $\{\hat{p}_j(0)\}$ for every $t > 0$.*

Physically speaking, this maximum principle states that the (rescaled) $\hat{p}_j(t)$ will not spontaneously form

localized regions with high probabilities of finding an adatom.⁶ For our purposes, it implies the following corollary:

Corollary. If $p_j(0) \leq \mathcal{O}(k/Z)$ for $j \neq s_0$ and $p_{s_0}(0) = \mathcal{O}(1/Z)$, then $p_j(t) \leq \mathcal{O}(k/Z)$ for $j \neq s_0$ and $p_{s_0}(t) = \mathcal{O}(1/Z)$ for all times t .

In essence, the corollary allows us to identify $p_{s_0}(t) = 1 - \mathcal{O}(kN)$ in Eq. (20) for all times, provided that $p_j(0) = \mathcal{O}(k/Z)$ for $j \neq s_0$ and $kN \ll 1$ [recall $Z = 1 + (N-1)k$]. Physically, the corollary states that if the system starts sufficiently close to equilibrium (i.e. the Boltzmann distribution), it will remain so for all times. The additional constraint $kN \ll 1$ implies a low total number of adatoms on the surface for all times. We henceforth refer to the hypotheses of the corollary as near-equilibrium conditions.

VI. CONTINUUM LIMIT AND THE LOW-DENSITY REGIME

In order to take the continuum limit of Eqs. (15), (16), and (20), we first identify the macroscopic parameters

$$\mathcal{D} = a^2 D, \quad (21)$$

$$c^{\text{eq}} = k/a, \quad (22)$$

$$\kappa_{\pm} = aD\phi_{\pm}. \quad (23)$$

We also enforce near-equilibrium conditions on the $p_j(0)$ and require $kN \ll 1$. Next, we assume that finite differences can be approximated in terms of a continuous density $c(x)$ via

$$\frac{c_{j+1}(t) - c_j(t)}{a} = \partial_x c(x, t)|_{x=ja} + \mathcal{O}(a), \quad (24)$$

$$\frac{c_{j+1}(t) - 2c_j(t) + c_{j-1}(t)}{a^2} = \partial_{xx} c(x, t)|_{x=ja} + \mathcal{O}(a). \quad (25)$$

Neglecting corrections that are $\mathcal{O}(a)$ or $\mathcal{O}[(kN)^2]$, we find that Eq. (15) reduces to the continuum step velocity law [Eq. (4)], Eq. (16) reduces to a diffusion equation [Eq. (1)], and Eq. (20) reduces to the linear kinetic relation at a step edge [Eq. (2)].

Because of the identity $\kappa_{\pm} = aD\phi_{\pm}$, the boundary conditions at the step edge depend strongly on the behavior of ϕ_{\pm} as $N \rightarrow \infty$. In the case that $\phi_{\pm} = \mathcal{O}(1)$ as $N \rightarrow \infty$, $\kappa_{\pm} \rightarrow \infty$, which forces the Dirichlet boundary condition $c^{\pm} = c^{\text{eq}}$; physically, this boundary condition corresponds to *diffusion-limited kinetics*, in which adatom hopping (as opposed to attachment/detachment) is the rate limiting process for the system to reach equilibrium. If, however, $\phi_{\pm} = \mathcal{O}(1/N)$ as $N \rightarrow \infty$, κ_{\pm} remains finite, and Eq. (2) gives the appropriate boundary conditions at the step edge.

A key benefit of this limiting procedure is the identification of the parameters entering the BCF theory (\mathcal{D} , κ_{\pm} , and c^{eq}) with the parameters of the microscopic model (D , k , ϕ_{\pm} , and a). It is possible to test the correctness

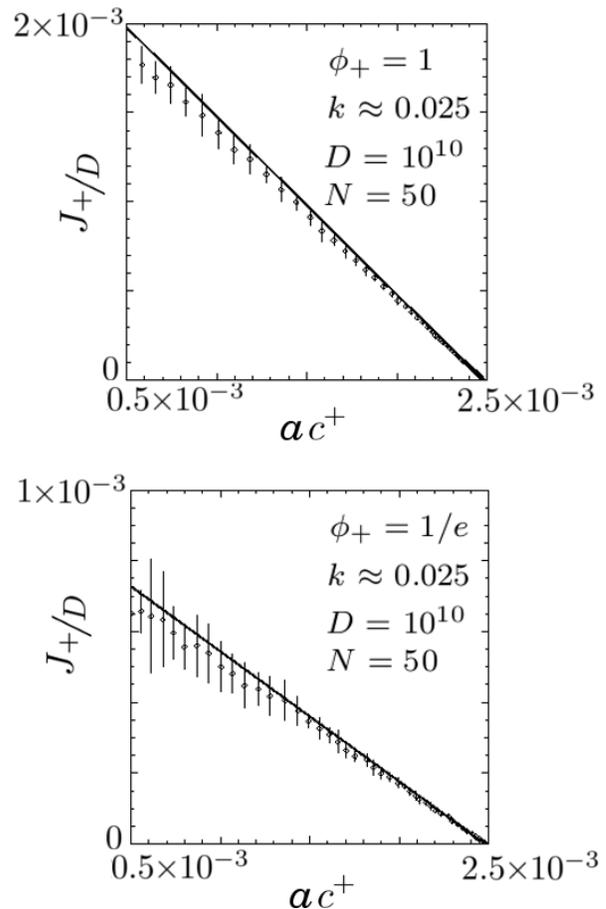


FIG. 4. Comparison of linear kinetic relation (2) (solid line) and kinetic Monte Carlo simulations (data points with error bars). (Top) Simulations with $\phi_+ = 1$, i.e. $E_+ = 0$. (Bottom) Simulations with $\phi_+ = 1/e$, i.e. $E_+/k_B T = 1$. We take the mean of 10 ensemble averages, with each ensemble consisting of: (a) 10^7 simulations, and (b) 10^6 simulations. The 3σ values are indicated by vertical lines (error bars) centered at the mean flux values. In both plots, we take $k \approx 0.025$, $D = 10^{10} \text{ s}^{-1}$, and $N = 50$ (so $a = L/50$); note that $ac^{\text{eq}} = k$. The slopes of the solid lines are (a) $J_+/[D(ac^+ - k)] = -1$ and (b) $J_+/[D(ac^+ - k)] = -1/e$, in agreement with our BCF-type model. See Sec. VI for discussion.

of these relations via kinetic Monte Carlo simulations. The main idea of such simulations is to follow many (10^6 or more) elements of the statistical ensemble describing the system and then compute ensemble averages with respect to those elements; assuming one samples enough elements of the ensemble, the simulated averages should approximate the true ensemble averages.

In Fig. 4 we compare our predictions of the parameters in the linear kinetic relation (solid line) versus kMC simulations (points) whose input parameters are the same as the atomistic SLG model. *Importantly, our BCF-type model has zero free parameters, since Eqs. (21)–(23) are determined entirely by the microscopic parameters of the kMC simulations.* The figures show that when

$c \approx c^{\text{eq}}$, the linear kinetic relation $\mathcal{J}_{\pm} = \mp \kappa_{\pm}(c^{\pm} - c^{\text{eq}}) = \mp(aD\phi_{\pm})[c^{\pm} - k/a]$ does a remarkable job of describing the atomistic behavior of the system.

VII. DISCUSSION

A. Derivation in the context of real materials

The derivation of our BCF-type model depends on several assumptions about the scaling of the microscopic parameters: (i) $kN \ll 1$ (ii) $D = \mathcal{O}(N^2) \text{ s}^{-1}$, and (iii) $N \gg 1$. In this section, we consider the validity of these assumptions in the context of real materials.

The hopping rate D is usually defined as the Arrhenius function $D = \tau^{-1}e^{-E_h/k_B T}$, where $\tau^{-1} = 10^{13} \text{ s}^{-1}$ is an attempt frequency and E_h is an activation energy that is extracted from measurements.¹⁸ Typical values for E_h range from 0.04 eV for Al(111) to 0.97 ± 0.07 eV for Si(111).¹⁸ At temperatures between 300 K and 1000 K, we estimate that $10^{12} \text{ s}^{-1} \geq D \geq 10^6 \text{ s}^{-1}$, depending on the material. As an example, we consider Ni(110), for which $E_h = 0.41 \text{ eV}$ ^{18,21}; taking $T \approx 500 \text{ K}$ (or $k_B T \approx 1/24 \text{ eV}$), we estimate that $D = 10^8 \text{ s}^{-1}$. For a terrace with $N = 1000$ lattice sites and $L = 0.1 \mu\text{m}$ (i.e. atomic length $a \sim 0.1 \text{ nm}$), we find $\mathcal{D} = D/(a^2) = 1 \mu\text{m}^{-2} \text{ s}^{-1}$.

Experiments can also estimate the energy E_b [cf. Eq. (12)]. Typical values range from approximately 0.3 eV for Ni(110)²¹ up to 1 eV or 2 eV for Si(111)²²⁻²⁴. The use of the value $E_b = 0.3 \text{ eV}$ for Ni(110) [cf. Eq. (12)] yields $k \approx 10^{-4}$ at 500 K. By combining this result with the assumption that $N = 1000$ (corresponding to L that is a few hundred nanometers), we find that $Nk \approx 10^{-1}$, which suggests that the low-density approximation is reasonable for this system at 500 K. In addition to these formal estimates, both experimental and numerical results have verified that Ni(110) is in a low-density regime at this temperature; see Ref. 21. In this work, significant adatom detachment on Ni(110) only began when the temperature was raised above 650 K; at 900 K, simulations show that roughly 1.5% of the lattice sites are occupied by adatoms (see also Ref. 19).⁷

Experimental estimates of E_{\pm} are also available [cf. Eq. (11)]. Often (but not always) the Ehrlich-Schwobel barrier^{25,26} E_- is larger than the attachment barrier E_+ . See, e.g., Table 6 in Ref. 18 for a detailed list of attachment/detachment barriers. For Ni(110), one finds $E_- = 0.9 \text{ eV}$ and $E_+ \approx 0 \text{ eV}$, which implies $\phi_- \ll 1/N$ and $\phi_+ = 1$ at 500 K. In a BCF model for this system, we therefore expect that $\kappa_- \approx 0$ and $\kappa_+ = \mathcal{O}(N)$, corresponding to $\mathcal{J}_- = 0$ and $c_+ = c^{\text{eq}}$. Therefore, for this system, our analysis predicts different boundary conditions on the two sides of the step edge.

B. Corrections due to multiple-particle states

The SLG master equation that we invoke in Sec. III only allows a single adatom to ever be on the surface. We comment here on corrections that can arise between multi-adatom states. Note that generally, the particular form of corrections will depend on precise rules of the multi-particle SLG model; here we only discuss only a few types. However, we always expect that an m -adatom state should be $\mathcal{O}[(kN)^m]$ (provided the system is near-equilibrium), since mE_b is the energy cost to create m adatoms. Thus, corrections from multi-particle states should always be small; see Ref. 13.

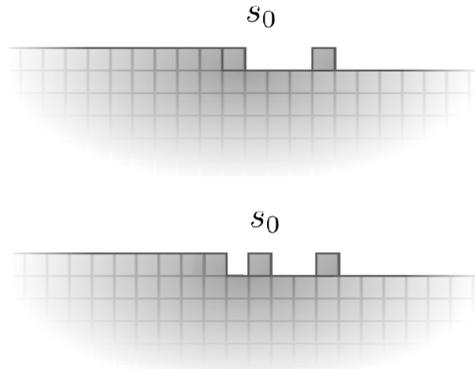


FIG. 5. A convection-type process that contributes corrections to the BCF theory. Here the adatom moved relative to the step because another atom detached from the step.

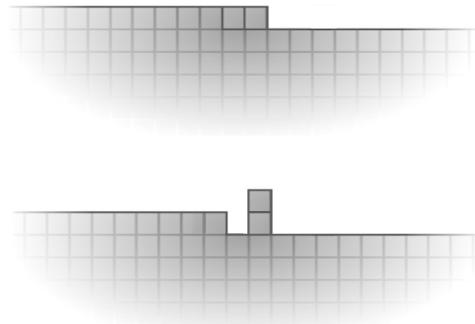


FIG. 6. Another process that yields corrections to the BCF theory: a transition between states in which the step changes its position by more than one lattice site.

In Fig. 5 transitions between the two states illustrated cause the adatom furthest from the step to move *relative to the step*. This is an example of a convection-type process. If the definition of the adatom density in a multi-particle model is taken relative to the step position, then such convective effects will appear in the diffusion equation for adatoms. Importantly, this process will not contribute a simple convection term $\zeta \partial_x c(x, t)$ to the diffusion equation (1) because the probabilities that *both* an

adatom is at some site *and* another atom detaches from the step are not independent; see Ref. 13.

In Fig. 6, transitions between the two states illustrated cause the step to move forwards or backwards by two lattice sites, as opposed to one. Any such process in which a single attachment (or detachment) event causes the step to move by more than one lattice site will introduce corrections to the step velocity law. Formally, this can be understood by examining Eq. (4): a single factor of a , as opposed to $2a, 3a, \dots$ etc., multiplies the net flux to the step. In the BCF theory, attachment of adatoms to a step causes it to move by a fixed amount.

C. Limitations

Our SLG model has limitations because we only consider a single step in 1D. In this setting, it is not possible to derive step interactions. In many formulations of the BCF theory, such interactions introduce an additional energy into the step chemical potential, so that the energy cost of adatom detachment depends on the widths of the terraces adjacent to the step.^{18,27–29} We speculate that in an appropriate multi-step SLG model, this energy penalty should appear as an additional, configuration-dependent contribution to E_b .

Because our SLG model is only 1D, we cannot account for the effects of anisotropy in the crystal lattice. Such effects could be important in systems such as Si(001), where diffusion rates depend on both direction and position.^{22,23} We speculate that an appropriate SLG model incorporating these features would lead to a BCF model with an anisotropic and (potentially) position dependent diffusion coefficient.

Our analysis is also unable to determine the role that kinks (i.e. bends in the step) play in the derivation of BCF-type models. In 2D SLG models, it is known that kinks, which alter the microscopic step profile, play an important role in determining the rates of adatom attachment/detachment processes.^{14,15} Moreover, in 2D BCF-type models, the chemical potential (i.e. the energy cost to remove an adatom from a step), and consequently the linear kinetic relations are typically assumed to depend on the local step curvature.²⁷ However, a derivation that expresses this dependence remains an open question.

VIII. CONCLUSIONS

In this paper, we showed how a 1+1D, single-step version of the BCF model can be derived from an atomistic, stochastic lattice-gas model of the surface. We used an

averaging procedure to connect the atomistic system configurations to the notions of a continuous adatom density and step position; we then showed that the BCF theory (plus corrections) describes the time evolution of these averages. Via a maximum principle, we also showed that corrections are negligible when the system is sufficiently close to equilibrium. Our use of an averaging procedure and maximum principle reveals the sense in which the BCF theory can be viewed as a model of mesoscale phenomena in non-equilibrium statistical mechanics.

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Appendix A: Proof of the Maximum Principle

Proposition. Let $p_j(t)$ be the solution to (5)–(9) with initial data $p_j(0)$, and define $\hat{p}_j = p_j/k$ for $j \neq s_0$ and $\hat{p}_{s_0} = p_{s_0}$. Let $\max_j\{\cdot\}$ be the greatest value in the set $\{\cdot\}$. Then \hat{p}_j satisfies the maximum principle that $\max_j\{\hat{p}_j(t)\} \leq \max_j\{\hat{p}_j(0)\}$ for all $t > 0$.

Proof. We proceed by *reductio ad absurdum*. Writing (5)–(9) in terms of \hat{p}_j yields

$$\begin{aligned} k\dot{\hat{p}}_j &= Dk[\hat{p}_{j+1} - 2\hat{p}_j + \hat{p}_{j-1}], & j \neq s_0, s_0 \pm 1, \\ k\dot{\hat{p}}_{s_0 \pm 1} &= Dk[\phi_{\pm}\hat{p}_{s_0} - (1 + \phi_{\pm})\hat{p}_{s_0 \pm 1} + \hat{p}_{s_0 \pm 2}], \\ \dot{\hat{p}}_{s_0} &= Dk[\phi_-\hat{p}_{s_0-1} - (\phi_- + \phi_+)\hat{p}_{s_0} + \phi_+\hat{p}_{s_0+1}]. \end{aligned} \quad (\text{A1})$$

Assume that at some time t there is an l such that $\dot{\hat{p}}_l(t) \geq 0$ and $\hat{p}_l(t) \geq \hat{p}_j(t)$ for all $j \neq l$. But by virtue of Eq. (A1),

$$\hat{p}_l(t) \geq \frac{\theta_1\hat{p}_{l-1}(t) + \theta_2\hat{p}_{l+1}(t)}{\theta_1 + \theta_2},$$

where $\theta_{1,2}$ are 1 or ϕ_{\pm} , depending on the value of l . By assumption, it is impossible that $\hat{p}_{l\pm 1}(t) > \hat{p}_l(t)$, so that either \hat{p}_l is not a maximum or \hat{p}_j is constant for all j .

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¹ For a more complete treatment of this derivation, see

- Ref. 13.
- ² The original BCF formulation¹⁰ amounts to $\kappa_{\pm} \rightarrow \infty$, so that $c = c^{\text{eq}}$ at the step edge.
 - ³ The actual value of s_0 is not important for our derivation. In fact, s_0 does not appear in the final form of the step velocity law.
 - ⁴ Later we will define the step position in the BCF theory as the expected microscopic step position.
 - ⁵ The barrier for an adatom to attach to the step from the upper terrace (in this case, E_-) is referred to as the Ehrlich-Schwoebel barrier.
 - ⁶ Maximum principles are often invoked in the analysis of diffusion equations of the form $\partial_t c(x, t) = D \partial_x^2 c(x, t)$, where c could be an adatom concentration or temperature, for example. In the latter case, the maximum principle states that the system does not spontaneously form localized hot spots.
 - ⁷ Nickel is in fact a more complicated system than our estimates lead one to believe; specifically diffusion is anisotropic on the (110) facet, and our simple BCF-type theory might not capture the subtleties in adatom motion due to this anisotropy. However, the energies we quote can be thought of as representative values since they lie roughly in the middle of a range seen in typical systems. See, for example, Ref. 18.
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